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## Synthesis of 3-(4-heteroaryl-phenyl)-8-oxabicyclo[3.2.1]octane-2-carboxylic acid methyl esters

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Abstract—Cross coupling protocols were applied for the synthesis of 3-(4-heteroaryl-phenyl)-8-oxabicyclo[3.2.1]oct-2-ene-2-carboxylic acid methyl esters. Stille conditions produced the corresponding products in reasonable yields. Samarium iodide reduction of the resulting coupling products produced the  $2\beta$ -carbomethoxy-3 $\alpha$ -aryl-8-oxabicyclo[3.2.1]octane diastereoisomers as the major, and the 2b-carbomethoxy-3b-aryl-8-oxabicyclo[3.2.1]octane diastereoisomer as the minor products. Both diastereomers manifested inhibition of the dopamine (DAT) and serotonin (SERT) transporters, with some selectivity for SERT inhibition. 2005 Elsevier Ltd. All rights reserved.

The discovery and development of potential medications for cocaine abuse has focused to a large extent on the diverse family of bicyclo<sup>[3.2.1]</sup>octanes.<sup>1-3</sup> These compounds bind potently to monoamine uptake mechanisms in the mammalian system (dopamine (DAT), serotonin (SERT) and norepinephrine (NET) transporters) and these systems have been implicated in the pharmacology of addiction caused by cocaine, itself a bicyclo[3.2.1]octane. In particular, the stimulatory and addictive properties of cocaine have been thought to be largely due to its ability to inhibit the  $\overline{DATA}^{4-9}$ although the SERT has also been implicated.<sup>10–13</sup>

In an extensive investigation aimed at developing medications for cocaine abuse, we have shown that the DAT and SERT inhibitory potency of the bicyclo[3.2.1]octane system is not limited by the nature of the heteroatom at the 8-position. While cocaine and its potent 3-aryl congeners 1 [\(Fig. 1\)](#page-1-0) are 8-azabicyclo[3.2.1]octanes, we have demonstrated that 8-carba,<sup>[14](#page-4-0)</sup> 8-oxa<sup>14</sup> and 8-thia<sup>[15](#page-4-0)</sup> bicyclo[3.2.1]octanes (2–4) have significant potency in inhibition of monoamine transport and are often equipotent to their 8-aza counterparts.<sup>[1](#page-3-0)</sup> In our search for SERT selective uptake inhibitors we have now explored the introduction of heteroaryl moieties on the 3-aryl ring within the 8-oxabicyclooctane series. In this regard, Davies et al.<sup>[16](#page-4-0)</sup> and Fu et al.<sup>[17](#page-4-0)</sup> have reported the synthesis and biological effects of 3-heterobiaryl systems in the 8-azabicyclo[3.2.1]octane series. They found that the introduction of 4- $(2$ -pyrrolyl)phenyl<sup>16</sup> and 4- $(2$ -thiophenyl)phenyl[17](#page-4-0) caused these molecules to increase binding potency at SERT and decrease binding potency at DAT, thus providing compounds in which the DAT vs. SERT selectivity of the monoaryl compounds had been reversed. We now present the synthesis of a series of 3-(4-heteroarylphenyl)-8-oxa-2-carbomethoxybicyclo- [3.2.1]octanes.

The synthesis of 8-oxabicyclooctanes such as 3 has been described in detail previously.[18](#page-4-0) The route that we had developed utilized a Suzuki coupling of the vinyltriflate 24 ([Scheme 2\)](#page-1-0) with appropriate arylboronic acids. The 4-bromophenyl compound [\(Scheme 1\)](#page-1-0) 5 was available via this route, so we initially planned to utilize 5 in a second Suzuki coupling to obtain 10–12. To our surprise, Suzuki conditions (Pd(PPh<sub>3</sub>)<sub>4</sub>, LiCl, Na<sub>2</sub>CO<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>) did not produce any product with 2-furanboronic acid, 6. Furthermore, only a low yield (36%) of 11 was obtained with 7 under these conditions. However, Stille coupling of 5 with both 8 and 9 proved successful (56–  $61\%$ ).

A more general synthesis of the bicyclooctanes 10–13 was developed and is shown in [Scheme 2.](#page-1-0) Since 24[18](#page-4-0) is readily available in our laboratories, we elected to use

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<span id="page-1-0"></span>

Figure 1.



 $Pd(PPh<sub>3</sub>)<sub>4</sub>$ , Dioxane, reflux





Scheme 2. Synthesis of biaryl-8-oxatropenes 10–13.

Stille coupling with appropriate stannanes to obtain the target compounds. Thus coupling of the arylstannanes 8, 9, 14, 15 with 1-bromo-4-iodobenzene, with  $Pd(PPh<sub>3</sub>)<sub>4</sub>$  as catalyst, provided the biaryl compounds 16–19, which were purified by column chromatography (40–87%). Treatment with *n*-BuLi at  $-70$  °C in THF for 1 h, followed by addition of tributyltin chloride then provided the tributylstannyl compounds 20–23. Attempts to purify these stannyl compounds resulted in substantial decomposition. Consequently they were used in the following reactions without purification.

The bicyclooxatropenes 10–13 were then obtained by a second Stille coupling. Thus cross coupling of the triflate  $24^{18}$  $24^{18}$  $24^{18}$  with the arylstannane building blocks  $20-23$  was carried out at room temperature in 1-methyl-2-pyrrolidinone (NMP) in the presence of  $Pd_2dba_3$ , tris-(2furyl)phosphine and  $ZnCl<sub>2</sub>$ . Treatment of the crude reaction mixture with 10% aq KF greatly simplified the removal of tin byproduct by transforming it into insoluble tributyltin fluoride, which can be removed by filtration. The products were obtained in high yields (75– 82%) after flash column chromatography on silica gel with ethyl acetate/hexane (1:9) as the eluent. When necessary, products were further purified by recrystallization. Characterization was performed by  ${}^{1}\text{H}$  NMR spectroscopy, mass spectrometry and elemental analysis. The <sup>1</sup>H NMR spectra of these compounds are diagnostic (Table 1). In particular the  $H_1$  proton appears as a doublet between  $\delta$  4.96–5.02 while H<sub>5</sub> is a multiplet at  $\delta$  4.64–4.68. Furthermore, H<sub>4B</sub> appears as a double doublet at  $\delta$  2.95–3.08 with the H<sub>4 $\alpha$ </sub> proton as a doublet at  $\delta$  2.14–2.28.

Earlier structure–activity relationship studies have pointed conclusively to the fact that the biologically interesting isomers within the broad class of bicyclo- [3.2.1]octanes have certain stereospecific requirements.<sup>1,18,19</sup> In particular, the 2 $\beta$ -carbomethoxy-3 $\beta$ -aryl (chair) configured compounds generally manifested potency at both DAT and SERT. In contrast, the  $2\beta$ -carbomethoxy-3 $\alpha$ -aryl (boat) compounds were generally far less potent at SERT. This lesser potency at SERT was also the case in the 2,3-unsaturated class of compounds to which  $10-13$  belong.<sup>[20](#page-4-0)</sup> For these reasons we were particularly interested in obtaining the  $2\beta$ -carbomethoxy-3b-aryl (chair) configured compounds. We anticipated that these would be inherently more potent at inhibition of the SERT, and in this new class of compounds that potency at SERT may be further enhanced by the presence of the new heteroaryl substituent. Reduction of the 2,3-unsaturated compounds 10–13 can be affected

Table 1. Physical and spectral data<sup>a</sup>

in many ways. However, methods such as catalytic hydrogenation result in reduction occurring from the  $\beta$ -face, thus resulting in  $2\alpha$ ,  $3\alpha$ -configured compounds. The most effective manner in which to obtain both the boat and chair 2b-carbomethoxy configured compounds has proved to be by the use of the single electron transfer reducing agent  $\text{SmI}_{2}^{21-23}$  in a solvent such as methanol.[18,24](#page-4-0) Thus, samarium iodide reduction of 10–13 in 10% methanol in tetrahydrofuran at  $-78$  °C provided the  $2\beta$ -carbomethoxy-3 $\alpha$ -aryl (boat) compounds in reasonable yield with the exception of 25d for which only a  $9\%$  yield was achieved (Scheme 3). The 2 $\beta$ -carbomethoxy-3b-aryl (chair) diastereomers were obtained in substantially lower yields for 26a and 26c while 26b and 26d were not obtained. In the latter case, the major product proved to be the  $2\alpha$ -COOCH<sub>3</sub>,  $3\alpha$ -aryl boat configured compound. Unfortunately the  $2\alpha$ -COOCH<sub>3</sub> diastereomers are often biologically uninteresting since they generally bind poorly at monoamine uptake sys-tems.<sup>[1,19](#page-3-0)</sup> Separation was achieved by column chromatography on silica gel using 10% ethyl acetate in hexane. Yields are given in [Table 2.](#page-3-0) Again, <sup>1</sup>H NMR spectroscopy was diagnostic for these compounds. The stereochemistry at the 2- and the 3-positions was established by the characteristic coupling constants and chemical shifts of the corresponding protons (Table 1). These are in good agreement with published work from our group.<sup>[18](#page-4-0)</sup> Specifically,  $3\alpha$ -aryl compounds manifest a double double doublet for H<sub>4x</sub> at about  $\delta$  1.4 whereas 3β-aryl compounds present a multiplet at  $δ$  1.9 for this proton. The proton at  $H_2$  is also diagnostic in that  $3\alpha$ aryl compounds manifest a double doublet at about  $\delta$ 2.6 whereas  $3\beta$ -aryl compounds have doublet at about  $\delta$  2.8.



Doublet (d), multiplet (m), double doublet (dd), double double doublet (ddd).

<sup>a</sup> Spectra measured in CDCl<sub>3</sub>.



Scheme 3. SmI<sub>2</sub> reduction of 3-biaryl-2-carbomethoxybicyclo<sup>[3.2.1</sup>]oct-2-enes.

	O $CO2CH3$ = -R 25			O CO <sub>2</sub> CH <sub>3</sub> 26	-R
Compound	$\mathbf R$	Yield (%)	Compound	$\mathbf R$	Yield (%)
25a		53	26a		21
25 <sub>b</sub>		79	26 <sub>b</sub>	S	Not obtained
25c	CH <sub>3</sub>	54	26c	CH <sub>3</sub>	22
25d		9	26d	S	Not obtained <sup>a</sup>

<span id="page-3-0"></span>Table 2. Distribution of diastereomers obtained from samarium iodide reduction

<sup>a</sup> The 2 $\alpha$ -COOCH<sub>3</sub>-3 $\alpha$ -aryl boat configured compound was the major product (25%).

It is noteworthy that within the series of 8-oxabicyclo<sup>[3.2.1]</sup>octanes, the  $3\alpha$ -aryl compounds are generally formed in higher yields than the  $3\beta$ -aryl compounds.<sup>[18](#page-4-0)</sup> Indeed, in the case of the reduction of 11 and 13, the  $3\beta$ -aryl compound 26b and 26d were not isolated at all. Only the 3 $\alpha$ -aryl 25b was obtained (79%), while 13 yielded a major product in which the  $2$ -COOCH<sub>3</sub> was isomerized to the  $2\alpha$ -configuration. The reasons for the general dominance of the  $3\alpha$ -aryl over the  $3\beta$ -aryl are unclear. Possibly, a preference for 3a-aryl reduced products might be dominated by an intermediate species in which the samarium is coordinated to the bridge oxygen, and probably to the C2-carbomethoxy group. It may then be surmised that proton transfer occurs preferentially from the  $\beta$ -face of the molecule to provide, predominantly, the 3a-aryl substituent.

The  $IC_{50}$  values for inhibition of the dopamine transporter (DAT) and the serotonin transporter (SERT are shown in Table 3). It is noteworthy that the reduced compounds 25 and 26 were generally more potent inhib-

Table 3. Inhibition of  $[3H]$ WIN 35,428 binding to the human dopamine transporter (hDAT) and [<sup>3</sup>H]citalopram binding to the human serotonin transporter (hSERT)<sup>a</sup>

		Compound DAT $IC_{50}$ (nM) SERT $IC_{50}$ (nM)	DAT/SERT <sup>b</sup>
10	102	248	0.4
11	272	242	1.1
12	>10,000	>10,000	1
13	>5,000	>14,000	1
25a	139	32	4.3
25 <sub>b</sub>	356	35	10.2
25c	1000	395	2.5
25d	1770	128	13.8
26a	64	30	2.1
26c	12,400	566	21.9

<sup>a</sup> Compounds are racemic. Each value is the mean of two or more independent experiments, each conducted in triplicate.

<sup>b</sup> Ratio of SERT inhibition to DAT inhibition.

itors of the SERT. However, in contrast to the parent 8 oxatropanes,[18](#page-4-0) both the boat and chair compounds manifested some selectivity versus inhibition of the DAT.

A series of (R,S) 3-(4-heteroarylphenyl)-8-oxabicyclo- [3.2.1]carboxylic acid methyl esters were synthesized via Stille coupling. Samarium iodide reduction of the bicyclooctenes produced the 3a-aryl diastereomer predominantly over the 3b-aryl diastereomer. Both 3a-aryl and the 3b-aryl diastereomers manifest a preferential inhibition of the serotonin transporter compared with their inhibition of the dopamine transporter.

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## Supplementary data

The supplementary data, including full experimental details, are available. Supplementary data associated with this article can be found, in the online version, at [doi:10.1016/j.tetlet.2005.10.170](http://dx.doi.org/10.1016/j.tetlet.2005.10.170).

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